

## Epitaxial Insulating Fluoride Layers on Semiconductors

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Received December 4, 1992; accepted December 23, 1992

IN HONOR OF SIR JOHN MEURIG THOMAS ON HIS 60TH BIRTHDAY

The epitaxial growth of the rare earth fluorides  $\text{HoF}_3$  and  $\text{GdF}_3$  on Ge, Si, and GaAs has been investigated. At the same time the detailed nature of the interfaces formed has been probed by X-ray photoelectron spectroscopy. Epitaxial layers can readily be grown and it is shown that layers with the high temperature tysonite-like phase may be prepared in such a way as to be stable at room temperature. A critical thickness exists where the layer structure changes to a multidomain orthorhombic form; this thickness is dependent on the degree of mismatch of the lattices of overlayer and substrate. Whereas the  $\text{GdF}_3$ -Ge interface is unreactive, strong reactions are seen for  $\text{HoF}_3$ -Si,  $\text{GdF}_3$ -GaAs, and  $\text{LaF}_3$ -Si. Reactions at the latter interface may be attenuated by an intermediate ordered layer of arsenic just one monolayer thick. © 1993 Academic Press, Inc.

### Introduction

There exists significant interest in the possibility of fabricating device structures such as integrated circuits based on a three dimensional configuration. The fabrication of such structures would be facilitated by the availability of semiconductor-insulator-metal multilayer structures. This possibility has led to an interest in the growth of epitaxial single crystal insulating fluoride layers on semiconductors. Most investigations have concentrated on the alkaline earth difluorides or mixtures of them grown in a lattice matched form on Si and III-V semiconductor (1, 2). These fluorides, however, have a rather large thermal mismatch (3) with most common semiconductors and are also slightly water soluble (4) which can be undesirable due to degradation of the layers in humid atmospheres.

The rare earth trifluorides, such as  $\text{LaF}_3$ , are not water soluble under normal conditions (4) and have a thermal expansion coef-

ficient closer to those of the group IV and III-V semiconductors (3).  $\text{LaF}_3$  has a trigonal, tysonite-like structure (5, 6) at all temperatures and is in fact reasonably matched to InP. However, the rare earth trifluorides ( $\text{REF}_3$ ) that might be more closely matched to Si, Ge, and GaAs possess an orthorhombic rather than a trigonal structure at normal temperatures (7), so that the question of epitaxy and lattice matching might at first appear irrelevant. However, closer inspection of these structures reveal that, over a narrow temperature range, solids such as  $\text{HoF}_3$  and  $\text{GdF}_3$  do in fact have a stable trigonal structure (7). This is shown in Fig. 1 where it may be seen that  $\text{HoF}_3$  and  $\text{GdF}_3$  appear to have stable tysonite structures above  $\sim 1000^\circ\text{C}$ . Several interesting questions arise from this result, two of which will be addressed in this paper.

(i) Can layers of  $\text{HoF}_3$  and  $\text{GdF}_3$  be grown in such a way that the trigonal tysonite structure is retained, even to room temperature, by growth on an appropriate substrate?

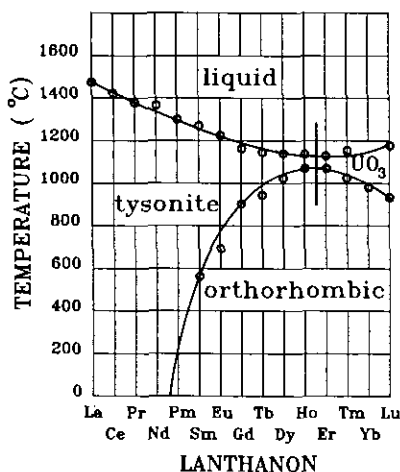


FIG. 1. Phase diagram showing the temperature dependent structural transitions for the  $RE\text{F}_3$ s. Data taken from (7).

(ii) For systems which are not lattice matched, can strained layers be grown and does a "critical thickness" exist as observed in the strained layer growth of common semiconductors?

It will be seen that the answer to both questions is yes. Also presented in the following are results and discussion pertaining to the general nature of some of the interfaces formed. The systems we have chosen are  $\text{HoF}_3$  on Si (111), where the lattice mismatch with the trigonal structure is around 2%, as well as  $\text{GdF}_3$  on Ge (mismatch 0.4%) and on GaAs (mismatch 0.4%). We briefly report also on the  $\text{LaF}_3$ -Si interface and in particular a method of reducing the interface reactions using an arsenic intermediate layer; abrupt and ordered interfaces are clearly desirable in any application.

## Experimental

Layers of the  $RE\text{F}_3$  were deposited in ultra-high vacuum from a simple tungsten heated spiral, the deposited thicknesses being measured using a quartz crystal thickness monitor. The ordering of the layers was studied with LEED and the composition and interface reactions by X-ray photoelectron

spectroscopy. Samples of Si and Ge were cleaned by heating to 1200 and 850°C respectively to produce (111) reconstructed surfaces, whereas GaAs (111)B clean surfaces were generated by heating arsenic capped MBE grown layers to drive away the cap.

## Results and Discussion

### (a) Structure of the Fluoride Overlayers

In general quite similar behavior was observed during the deposition of  $\text{HoF}_3$  on Si(111),  $\text{GdF}_3$  on Ge (111), and  $\text{GdF}_3$  on GaAs (111) surfaces. This is illustrated in Fig. 2 by references to  $\text{HoF}_3$  on Si (111). Here,  $\text{HoF}_3$  was deposited on room temperature substrates which were subsequently annealed, or else the samples were maintained at elevated temperatures during fluoride deposition with all subsequent measurements being at room temperature. Figure 2 shows the LEED pattern after the deposition of 12 Å of  $\text{HoF}_3$  at a substrate temperature of 750°C. This pattern is observed for  $\text{HoF}_3$  thicknesses in the range 1–12 Å and resembles the Si ( $\sqrt{3} \times \sqrt{3}$ ) $R30^\circ$  structure. For thickness of  $\text{HoF}_3$  in excess of 12 Å the LEED pattern changes, giving that shown in Fig. 3c. The interpretation of these patterns is considered below and is much aided by the diagrams of the  $\text{HoF}_3$  trigonal and orthorhombic structures shown in Fig. 4.

The  $\text{HoF}_3$  layer with thickness in excess of 12 Å is in fact the orthorhombic structure. The (001) projection shown in Fig. 4b yields a rectangular surface mesh in reciprocal space (Fig. 3a) and it is clear that domains of this structure exist, such that each domain is rotated by 120° with respect to the others. The LEED pattern is synthesized from the three rotated rectangular arrangements as shown in Figure 3b. In particular the "three spot" arrangement readily identifies this three domain orthorhombic phase. The LEED pattern for thicknesses below 12 Å also reflects the  $\text{HoF}_3$  structure but in this case it is the trigonal "tysonite" structure, illustrated in Fig. 4a, that is observed. It is

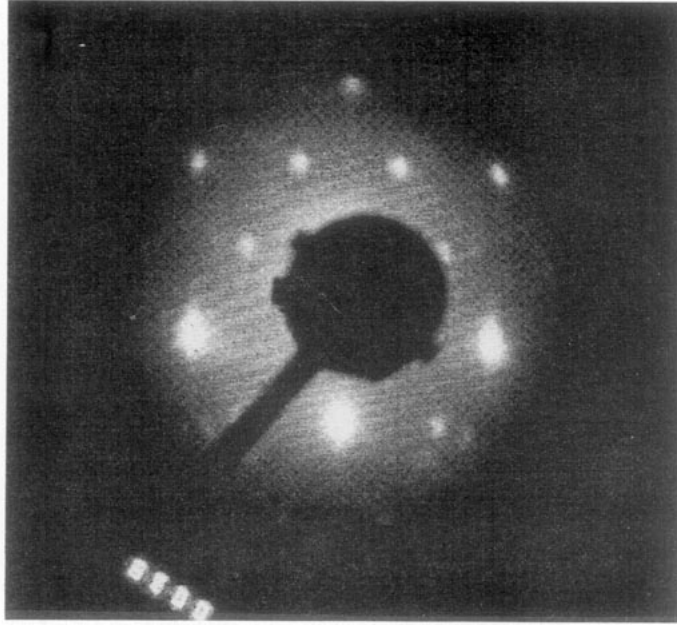


FIG. 2. LEED pattern recorded after  $\sim 12 \text{ \AA}$  of  $\text{HoF}_3$  had been deposited onto atomically clean Si (111) at  $750^\circ\text{C}$ .

apparent that up to  $12 \text{ \AA}$  the tysonite structure is indeed *stable at room temperature*. Beyond  $12 \text{ \AA}$  the growth mode also changes from layer upon layer to an island growth mechanism. The  $12 \text{ \AA}$  thus appears highly reminiscent of the critical thickness seen in the growth of strained layer semiconductors. The thickness is also of the order expected for a lattice mismatch of around 2%.

To further investigate the above possibilities similar experiments were carried out for trifluoride layers on Ge and GaAs. Here  $\text{GdF}_3$  was chosen in view of its close match (0.4% to Ge and GaAs). Here again the trigonal structure was found to be stable at room temperatures for thin layers, up to around  $180 \text{ \AA}$  for the Ge (111) substrates and  $350 \text{ \AA}$  for GaAs. Beyond  $200 \text{ \AA}$  the layers on Ge again display the multidomain orthorhombic structure. However, it is somewhat more complex than that for  $\text{HoF}_3$  on Si and will be considered in detail elsewhere (8). The critical thickness is substantially greater for  $\text{GdF}_3$  on Ge as expected. For  $\text{GdF}_3$  on

GaAs we have not so far observed the orthorhombic structure, but instead a disordered one, for thicknesses in excess of  $350 \text{ \AA}$ .

#### (b) Interface Reactions

The precise chemistry of the fluoride–semiconductor interface was monitored by detailed X-ray photoelectron spectroscopy measurements as a function of thickness and annealing cycles. Figures 5 and 6 show previously published spectra (9) for  $\text{HoF}_3$  on Si (111) for different temperatures and annealing times. Strong reactions are seen as evidenced by the large scale changes in the emission from the Si  $2p$  core level. Detailed analysis shows that a holmium silicide is formed, with the associated loss of fluorine from the interface. The thickness of the silicide layer increases with annealing, and its metallic character leads inevitably to the pinning of the Fermi level at the interface. Exceptionally strong interface reactions were seen for  $\text{GdF}_3$  on GaAs (111),

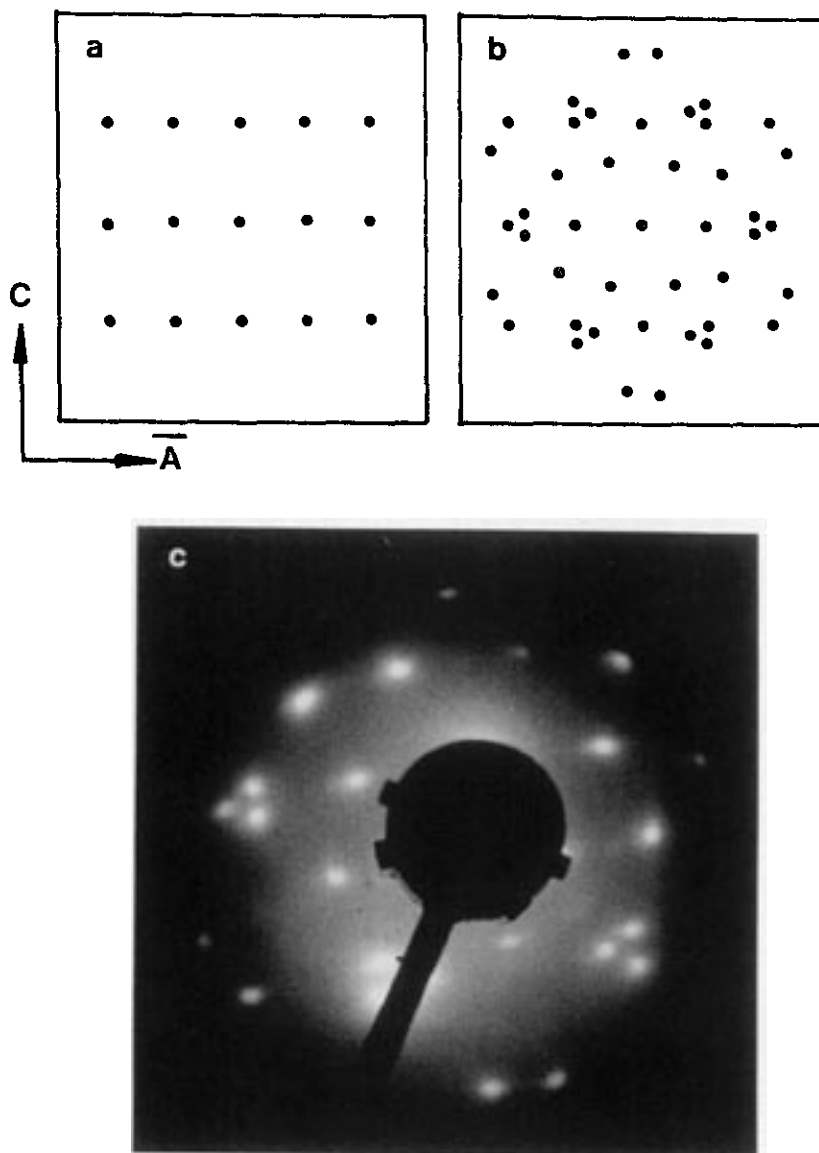


FIG. 3. (a) The reciprocal space mesh corresponding to orthorhombic  $\text{HoF}_3$  in the  $[010]$  direction. (b) The mesh with an additional two components rotated by  $120^\circ$  and  $240^\circ$  respectively. This reproduces the LEED pattern shown in (c) which was recorded with an incident beam energy of 36 eV.

leading to a range of reaction products. For  $\text{GdF}_3$  on Ge (111), however, we have so far not been able to detect the presence of interface reactions. Both of the above investigations will be described in full elsewhere (8, 10).

#### (c) Reaction Barriers at Interfaces

It is clear that the interface states associated with strong chemical reactions at the fluoride-Si and fluoride-GaAs interfaces are highly undesirable in any application. The suppression of reactions is thus an im-

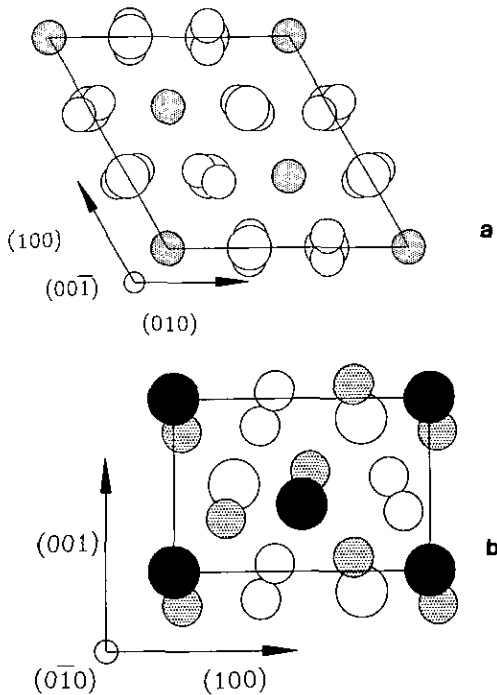


FIG. 4. (a) The unit cell of the orthorhombic structure of a  $\text{REF}_3$  viewed along (010). The unit of cell of the trigonal, tysonite-like structure of a  $\text{REF}_3$  viewed along (001).

portant issue. To see how this may be aided we consider the case of  $\text{LaF}_3$  on Si (111). In this case it is known that a lanthanum silicide layer is formed when the  $\text{LaF}_3$ -Si interface is annealed (11). This is apparent in the X-ray photoelectron spectra for  $\text{LaF}_3$  on Si shown in the upper (c) and lower (a) parts of Fig. 7, where the La 4d doublet displayed a pronounced shift to lower binding energy associated with the silicide formation. The  $\text{LaF}_3$  thick layer has the anticipated tysonite structure.

Figure 7 also shows spectra where the  $\text{LaF}_3$ -Si interface includes an arsenic interlayer (b). The As ordered monolayer was deposited on the Si (111) surface prior to the deposition of the  $\text{LaF}_3$ . It is clear that the form of the La 4d emissions changes very little upon annealing indicating significant suppression of the interface chemical reactions.

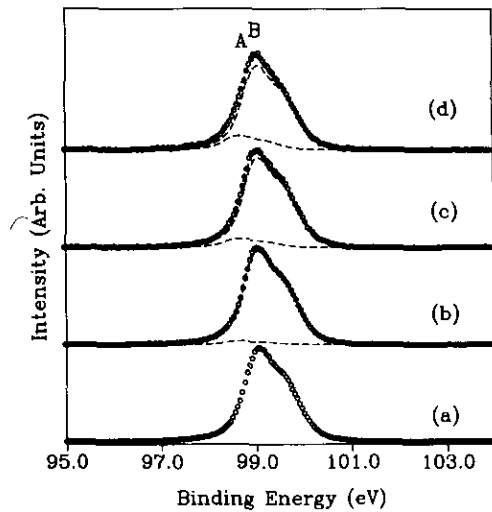


FIG. 5. Si 2p EDCs recorded (a) from the clean Si(111)- $7 \times 7$  substrate and after (b)-(d)  $\sim 7 \text{ \AA}$ ,  $12 \text{ \AA}$ , and  $41 \text{ \AA}$  respectively of  $\text{HoF}_3$  had been deposited at a substrate temperature of  $\sim 700^\circ\text{C}$ . Each spectrum has been scaled to the same height. Clearly, (b)-(d) can be synthesized with two components; B due to emission from bulk Si and A attributed to emission from Si bonded to Ho at the interface. A is offset from B by  $\sim 0.4 \text{ eV}$ .

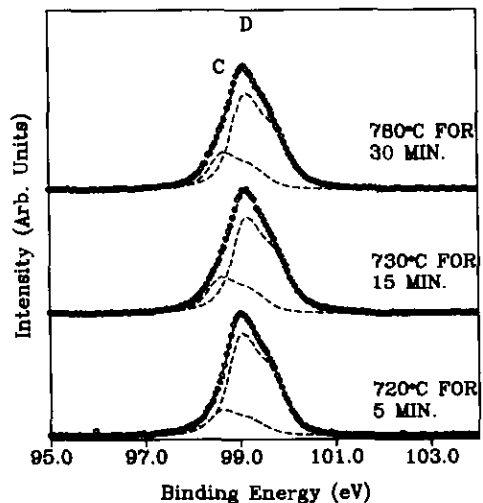


FIG. 6. Si 2p EDCs recorded as a function of annealing the  $41 \text{ \AA}$   $\text{HoF}_3$  layer on Si mentioned in connection with Fig. 5. Each spectra can be synthesized with two components; C shifted by  $\sim 0.5 \text{ eV}$  to the lower binding energy side of the bulk Si derived feature, D. Each spectrum has been scaled to the same height.

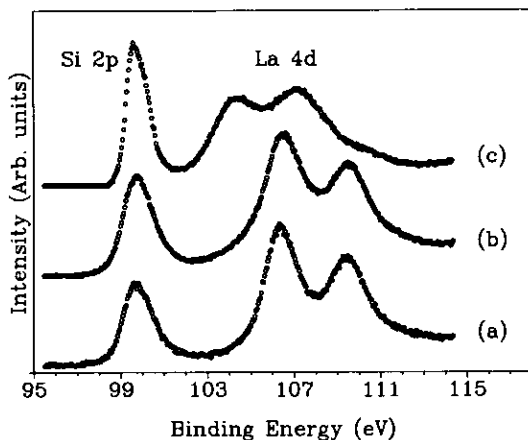


FIG. 7. (a) 10 Å of  $\text{LaF}_3$  deposited onto Si(111) grown at RT. (b) 10 Å of  $\text{LaF}_3/\text{As}/\text{Si}$  (111) annealed at 500°C for 30 min. (c) 10 Å of  $\text{LaF}_3$  on Si(111) annealed at 500°C for 30 min. Notice the large shift of the La 4d derived peak in (c) indicating the formation of a silicide. No reactions are evident in (a) or (b).

In conclusion we have shown that epitaxial insulating fluoride layers can readily be grown on group IV and III-V semiconductors. It is shown that structures normally stable only at high temperatures can be grown on Si and Ge such that the normally high temperature phase is stable at room temperature, thus considerably enhancing the range of suitable materials for epitaxial layers. The concept of a critical thickness, where the structures change from trigonal to orthorhombic is reminiscent of strained layers of group IV and III-V semiconductors. Interface reactions are common in the preparation of epitaxial layers and may lead to undesirable interface states. It is shown that chemical reactions can be substantially suppressed by the use of ordered monolayers deposited on the semiconductor be-

fore deposition and annealing of the fluoride layers. The potential of these findings is being further explored.

### Acknowledgments

One of us (RHW) expresses his sincere thanks to Sir John Meurig Thomas, FRS for introducing him to the technique of XPS and for his motivating influence over the years. LJ and AH acknowledge the U.K. Science and Engineering Research Council, and CG acknowledges the U.S. Air Force, for financial support.

### References

1. R. F. C. FARROW, P. W. SULLIVAN, G. M. WILLIAMS, G. R. JONES, AND D. G. CAMERON, *J. Vac. Sci. Technol.* **19**, 415 (1981).
2. P. W. SULLIVAN, R. F. C. FARROW, AND G. R. JONES, *J. Cryst. Growth* **60**, 403 (1982).
3. Y. S. TOULOUKIN (Ed.), "Thermophysical Properties of Matter," Plenum, New York (1977).
4. R. C. WEAST (Ed.), "Handbook of Chemistry and Physics," CRC Press, Boca Raton, FL (1985).
5. B. MAXIMOV AND H. SCHUTZ, *Acta Crystallogr. Sect. B* **41**, 88 (1984).
6. A. ZALKIN AND D. H. TEMPLETON, *Acta Crystallogr. Sect. B* **41**, 91 (1984).
7. R. E. THOMA AND G. D. BRUNTAN, *Inorg. Chem.* **1937** (1966). NB, however, that Paster and Robinson (*Mat. Res. Bull.* **9** 569 (1974)) showed that the high temperature transitions appear to be impurity sensitive ( $\text{OH}^-$  substituting  $\text{F}^-$ ) and are absent in the pure materials, with the exception of  $\text{SmF}_3$  and  $\text{EuF}_3$ .
8. L. C. JENKINS, C. L. GRIFFITHS, J. RICHARDS, AND R. H. WILLIAMS, in "Proceedings of the Fourth International Conference on the Formation of Semiconductor Interfaces, 14-18 June, 1993, Jülich, Fed. Rep. of Germany," to appear.
9. C. L. GRIFFITHS, J. E. MACDONALD, AND R. H. WILLIAMS, *Appl. Surf. Sci.* **56-58**, 782 (1992).
10. L. C. JENKINS AND R. H. WILLIAMS, to be published.
11. C. L. GRIFFITHS, Ph.D. thesis, University of Wales (1991).